CHM 421/621 Statistical Mechanics

Lecture 17 Alternative Formulations

Introduction and Review

Lecture Plan

Review of Thermodynamics

Basic Formalism

Conditions of Equilibrium

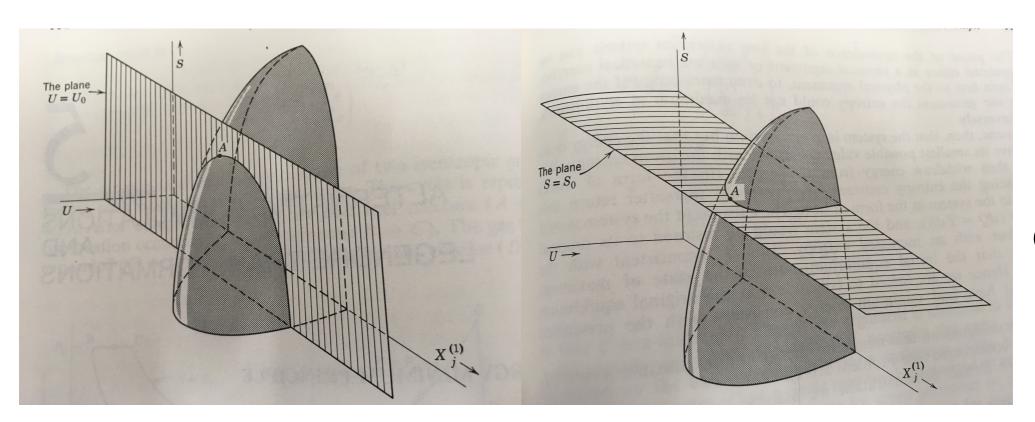
Equilibrium Relations

Legendre Transformed Representations

Stability of Thermodynamic Systems

Minimum energy principle

The equilibrium value of any unconstrained internal parameter is such as to minimise the energy for the given value of the total entropy.



Taken from Callen, Sec. 5-1

The properties of thee fundamental equation, i.e., the single-valuedness of U w.r.t. S and $\left(\frac{\partial S}{\partial \overline{S}}\right) > 0$ ensure that this can happen.

Minimum energy principle

Proof:

Consider a composite system. At a given energy *U* we have that

Entropy maximisation =>
$$\left(\frac{\partial S}{\partial X}\right)_U = 0$$
 and $\left(\frac{\partial^2 S}{\partial X^2}\right)_U < 0$

where we have denoted a generic extensive parameter for one of the sub-systems as *X* for simplicity of notation. We also take it to be implicit that all other parameters are held fixed in the derivative.

We further assume the following notation for the corresponding (energetic) conjugate variables

$$P \equiv \left(\frac{\partial U}{\partial X}\right)_S$$

Minimum energy principle

Proof:

Now,

$$P = \left(\frac{\partial U}{\partial X}\right)_S = -T\left(\frac{\partial S}{\partial X}\right)_U$$
 How?
$$= 0$$

=> U has an extremum at the same point X.

Now to classify the extremum as a maximum or a minimum ...

Minimum energy principle

Proof:

Let's calculate
$$\left(\frac{\partial^2 U}{\partial X^2}\right)_S$$

$$= \left(\frac{\partial P}{\partial X}\right)_{S}$$

$$= \left(\frac{\partial P}{\partial U}\right)_{X} \left(\frac{\partial U}{\partial X}\right)_{S} + \left(\frac{\partial P}{\partial X}\right)_{U}$$

$$= \left(\frac{\partial P}{\partial U}\right)_{Y} P + \left(\frac{\partial P}{\partial X}\right)_{U} = \left(\frac{\partial P}{\partial X}\right)_{U}$$

Thus, *U* is minimum!

$$=-T\frac{\partial^2 S}{\partial X^2}>0$$
 at $\frac{\partial S}{\partial X}=0$

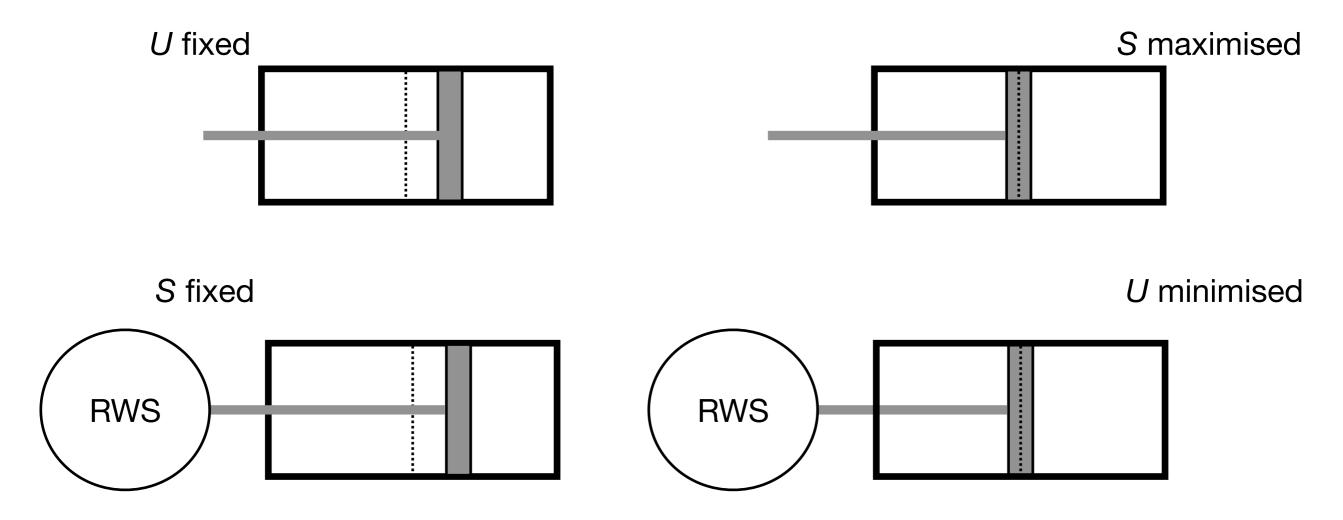
at
$$\frac{\partial S}{\partial X}=0$$

How?

At P = 0

Minimum energy principle

Illustration:



Can use this principle instead of entropy maximisation for solving problems.